

Poly- and Perfluoroalkyl Substances (PFASs): An Introduction

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Ground Water and Drinking Water

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Drinking Water Health Advisories for PFOA and PFOS

Health Advisories

EPA has established health advisories for PFOA and PFOS based on the agency's assessment of the latest peer-reviewed science to provide drinking water system operators, and state, tribal, and local officials who have the primary responsibility for overseeing public water systems, with information on the health risks of these chemicals. They can take the appropriate actions to protect their communities. EPA is committed to supporting states and public water utilities to determine the appropriate steps to reduce exposure to PFOA and PFOS in drinking water. As science on health effects of these chemicals evolves, EPA will continue to evaluate new information.

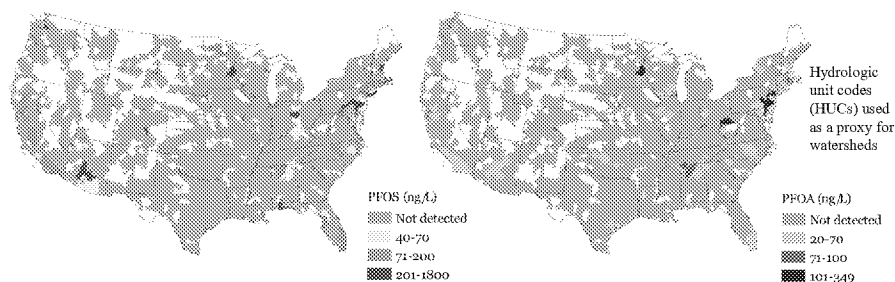
PFOA	70 ng/L
PFOS	70 ng/L
PFOA + PFOS	70 ng/L

To provide Americans, including the most sensitive populations, with information on the health risks of PFOA and PFOS from drinking water, EPA has established health advisory levels at 70 parts per trillion.

What's a health advisory?

non-enforceable and non-regulatory

A national issue?: UCMR 3 Data



- EPA survey of all public water systems (PWSs) serving 10,000+
- Chemicals monitored: PFBS, PFHxS, PFOS, PFHpA, PFOA, PFNA
- Through 2015, 193 PWSs (3.9%) had detectable PFAAs
- Fire training areas and aqueous film forming foam (AFFF) releases likely important sources



Hu *et al.*, 2016 ES&TL

Terminology and Acronyms



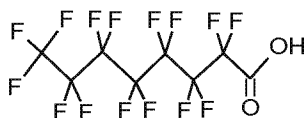
PFC = perfluorinated chemical, “perfluorochemical”

PFC = perfluorocarbons, a family of greenhouse gases

Poly- and perfluoroalkyl substances = PFASs

Perfluoroalkyl acids = PFAAs (a subclass of PFASs)

PFAAs and PFASs are not synonymous



Perfluoroalkyl acid



Polyfluoroalkyl acid

Buck et al., 2011, IEAM.

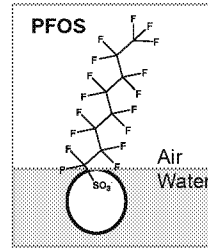
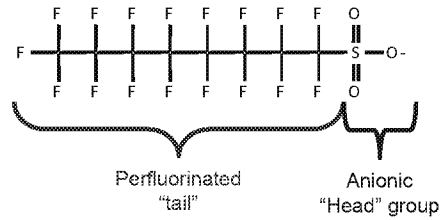
Perfluoroalkyl acids (PFAAs)



- Fully fluorinated chemicals that repel both oil and water
- PFAA-based precursors (i.e., ***poly***fluorinated substances) used in coatings for textiles and paper packaging products, fire-fighting foams, etc.
- Persistent, Bioaccumulative, and Toxic (PBT)
- Widely detected in wildlife and humans
- Relatively mobile and yet bioaccumulative

So, again, what are PFAAs?

Perfluorooctane
sulfonate (PFOS)
 $C_8F_{17}SO_3^-$



Good news: C-F bond is one of the strongest chemical bonds known

Bad news: C-F bond is one of the strongest chemical bonds known

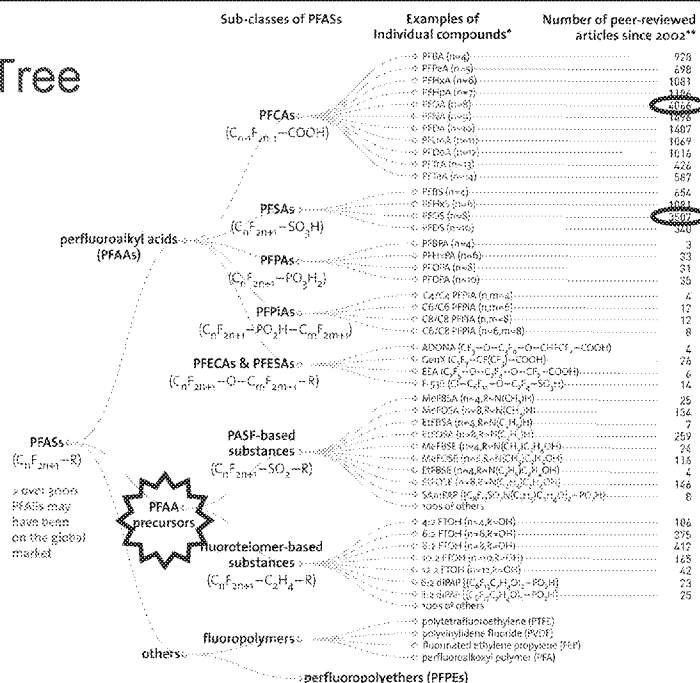
PFAAs are
extremely persistent
in the environment



PFAS Family Tree

It's not just
PFOS and
PFOA

Wang et al. 2017, ES&T.



PFAS Sources

- Consumer products
 - Wastewater, biosolids, and landfills
- AFFF releases
 - Not just military sites
- Chromium electroplating
 - PFASs used for mist suppression
 - PFCAs (C4-C11) and PFASs (C4,6,8) in discharge water¹
 - 6:2 FtS 'alternative' mist suppression agent²
- Industrial (plastics/polymer/textile) manufacturing sources
 - PFNA: West Deptford, NJ Solvay Specialty Polymers³
 - PFOA: Saint Gobain Performance Plastics and Honeywell polymer manufacturing in Hoosick Falls, NY⁴

¹EPA, 2009. PFOS Chromium Electroplater Study

²Yang et al., 2014. Environ Sci Pollut 21:4634-4642

³<http://www.njspotlight.com/stories/15/04/06/drinking-water-panel-calls-for-stricter-standard-on-potential-carcinogen/>

⁴<http://www.villageofhoosickfalls.com/news.html>

Slide content courtesy of ESTCP Project ER-201574-T2. Full FAQ presentation available at <https://youtu.be/lyzSc8FF752E>

Jennifer Field: End

Aqueous Film Forming Foam (AFFF)

- Only 3% of fluorochemical production is for AFFF
 - 75% of AFFF production used by military
 - 25% used by oil refineries, municipal airports & fire stations, tank farms
- Complex, proprietary mixtures
- PFASs a few % in mixture but still g/L levels
- Brief history
 - Mid 1960s – 1970: 3M sole source supplier of AFFF
 - 1973: National Foam
 - 1976: Ansul
 - 1994 to present: Angus, Chemguard, Fire Service Plus
- Multiple AFFFs used at most sites, **often directly released**

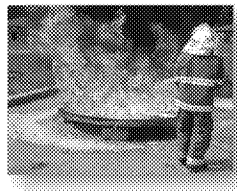


Photo courtesy of John Farley, Director, CBD/ex-USS SHADWELL Fire Test Operations, Naval Research Laboratory, Washington DC; 'Moody et al., 2000. *ES&T* 34: 3864-3870

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Consider 1,000 mg (1 g/L), when dissolved completely in water, will contaminate a lot of groundwater

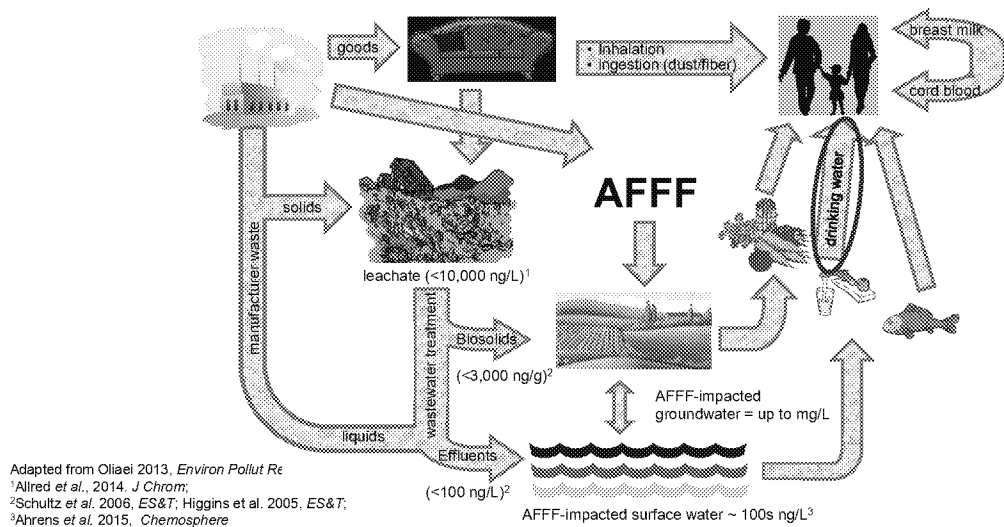
PERSONALITY – goes to interface to put out fire (< 1 min)

When we analyze, must dilute 1:1,000,000

We have to stop and think – 'only a few %' = parts per hundred

RITS 2016: Emerging Contaminants

Human Exposure to PFASs



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